analogue. The cross-linking of the thiazolium salt polymer catalyst was found to be useful in protecting an active thiazolium salt unit from attack by hydroxide ions at high pH, in the same way as thiamine pyrophosphate in vivo is protected by the surrounding apoenzyme. The polymer catalyst has higher activity than the low molecular weight analogue not only because of the protection of the active site but also because the hydrophobic substrate is preferentially adsorbed to the polymer domain by hydrophobic interaction.

References and Notes

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CORRECTIONS

Wesley Memeger, Jr.: Novel Thermotropic Main-Chain Polyhydrocarbons and Block Copoly(hydrocarbon–azomethines). Volume 22, Number 4, April 1989, p 1577.

1. The temperature in the legend to Figure 5 should read 260 instead of 270 °C.
2. The PMT of polymer D in Table I should read 220 instead of 300 °C.

V. Pavone, E. Benedetti,* V. Barone, B. Di Blasio, F. Lelj, C. Pedone, A. Santini, M. Crisma, G. M. Bonora, and C. Toniolo*: Structural Versatility of Peptides from $C^{\alpha,\alpha}$-Dialkylated Glycines. A Conformational Energy Computation and X-ray Diffraction Study of Homopeptides from 1-Aminocyclohexane-1-carboxylic Acid. Volume 21, Number 7, July 1988, p 2064.

In Table II, the bond distances $C^{\alpha}{}_i$–$C^{\beta 1}{}_i$, $C^{\alpha}{}_i$–$C^{\beta 2}{}_i$, $C^{\beta 1}{}_i$–$C^{\gamma 1}{}_i$, $C^{\beta 2}{}_i$–$C^{\gamma 2}{}_i$, $C^{\gamma 1}{}_i$–C^{δ}, and $C^{\gamma 2}{}_i$–C^{δ} given for *p*-BrBz(Acc^6)$_4$O-*t*-Bu, res 2, should be 1.560 (10), 1.546 (10), 1.360 (11), 1.358 (12), 1.540 (14), and 1.534 (13) Å, respectively.